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14 (CHM-2) CHM 205

2019

CHEMISTRY

Paper: CH 205

(Green Chemistry 2)

Full Marks: 40

Time: Two hours

The figures in the margin indicate full marks for the questions.

- 1. Answer any four questions: $4 \times 1 = 4$
 - (a) Which of the following is NOT a physical treatment of waste material?
 - (i) Filtration
 - (ii) Centrifugation
 - (iii) Precipitation
 - (iv) Absorption
 - (b) Define Environmental Quotient.

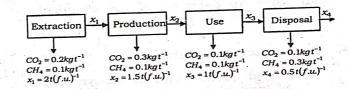
- (c) Turnover frequency of catalyst A is more than that of catalyst B. State which catalyst is more active.
- (d) Green chemistry is not different from traditional chemistry in as much as it, embraces the same _____, and ____ that has always been central to classical chemistry. (Fill up the gaps)
- (e) Wittig reaction is an atom economic reaction. 1
 (True/False)
- (f) Sustainability is a transition from _____ sources of energy to ____ one. (Fill up the gaps)
- 2. Answer any two questions: 2×3=6
 - (a) Define atom economy. Calculate the % atom economy for the following reaction 1+2

 $CH_3(CH_2)_2CH_2OH + SOCl_2 \rightarrow CH_3(CH_2)_2CH_2Cl + SO_2 + HCl$ 102 119 120.5

- (b) Using suitable examples, illustrate how green chemistry is synonymous with sustainable chemistry.
- (c) Comment on the usefulness of CO_2 as a solvent in organic synthesis.
- 3. Answer any two questions: 2×4=8

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(a) What is environmental burden? Calculate the total environmental burdens (in f.u.) due to the emissions of (i) CO₂ (ii) CH₄ in the system shown below:



(b) Compare aerobic and anaerobic waste treatment processes. Show how catechol is formed by aerobic treatment of methyl benzene. Why is the formation of catechol important in terms of green chemistry?

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(c) Taking DDT
(dichlorodiphenyltrichloroethane) —

a wonder chemical that gave its inventors a Nobel Prize - as an example, illustrate how environmental persistence of a chemical product must be considered as one of the criteria for designing new insecticides.

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- (d) Give an overview of the chemical recycling of polymers to monomers.
- 4. Answer any two questions:

2×4=8

- (a) Provide green routes for the synthesis of (any two) 2+2
 - (i) Caprolactum
 - (ii) Ibuprofen
 - (iii) Acetic acid
- (b) Discuss briefly the problems associated with the conventional synthesis of

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- phenol via benzene sulfonation. Provide an alternative method for synthesis of phenol from green chemistry perspective. 4
- (c) Write down the products of any two of the following reactions and highlight the greenness of the reactions: (any two)

4)
$$R^{NH} + R^{O} CI \xrightarrow{SiO_2}$$

- 5. Answer any two questions:
 - (a) Define biomass as a renewable raw material (RR.M) and list the major advantages and current disadvantages of utilizing plant biomass for meeting our need of materials and fuel.

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Contd.



- What are the major components of lignocellulosic biomass? Discuss the various ways in which it may be transformed into chemicals and fuels.
- Succinic acid, Comment on the green (c) chemistry advantages of utilizing biomass-derived succinic acid for producing many chemical products that are useful in the pharmaceutical, cosmetic and food industries, among other. Give one indicative reaction to convert succinic acid to a more useful chemical product.
- Answer any two questions:
 - What are the green credentials of photochemical processes in chemical transformations?

- What do you mean by "Sustainable Development"? Indntify the targets that should be attained for sustainable development under green chemistry perspective.
- Write a short-note on the sustainable use of water.

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